



# Certificate

Produced in double accredited laboratory fulfilling ISO/IEC 17025 and ISO 17034

This certificate is designed in accordance with ISO Guide 31 [1].

Object of certification: Multielement standard solution 1 for ICP

Product No.: **90243** Lot: **BCCB6994** 

Composition: High-purity starting materials in 10% HNO $_3$  (prepared with high purity water, 18.2 M $\Omega$ cm,

0.22 µm filtered, and acid suitable for trace analysis).

Density at 20°C:  $\rho = 1056.3 \text{ kg m}^{-3}$   $u_c(\rho) = 0.5 \text{ kg m}^{-3}$ 

Intended use: Calibration of ICP, AAS, spectrophotometry or any other analytical technique.

Storing and handling: This reference material shall be stored between 5°C and 30°C. The bottle's temperature must

be 20°C and shaken well before every use. If storage of a partially used bottle is necessary, the cap should be tightly sealed and the bottle should be stored at reduced temperature (e.g.

refrigerator) to minimize transpiration rate.

Expiry date: **MAY 2023**Certificate issue date: 17 JUL 2019

Certificate version: 01

Bottle opening date:

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Certified value traceable to SI unit kg and uncertainty according to ISO Guide 35 [2] and Eurachem/CITAC Guide [3]								
Constituent	Certified values at 20°C and expanded uncertainty [U = k·uc; k = 2]							
Aluminum (Al)	47.3	mg kg⁻¹	±	0.3	mg kg <sup>-1</sup>	50.0 mg L <sup>-1</sup>	±	0.3 mg L <sup>-1</sup>
Barium (Ba)	9.47	mg kg <sup>-1</sup>	±	0.03	mg kg <sup>-1</sup>	10.00 mg L <sup>-1</sup>	±	0.03 mg L <sup>-1</sup>
Bismuth (Bi)	94.7	mg kg <sup>-1</sup>	±	0.3	mg kg <sup>-1</sup>	100.0 mg L <sup>-1</sup>	±	0.3 mg L <sup>-1</sup>
Boron (B)	47.3	mg kg <sup>-1</sup>	±	0.1	mg kg <sup>-1</sup>	50.0 mg L <sup>-1</sup>	±	0.1 mg L <sup>-1</sup>
Cadmium (Cd)	9.47	mg kg <sup>-1</sup>	±	0.03	mg kg <sup>-1</sup>	10.00 mg L <sup>-1</sup>	±	0.03 mg L <sup>-1</sup>
Calcium (Ca)	9.47	mg kg <sup>-1</sup>	±	0.03	mg kg <sup>-1</sup>	10.00 mg L <sup>-1</sup>	±	0.03 mg L <sup>-1</sup>
Chromium (Cr)	47.3	mg kg <sup>-1</sup>	±	0.2	mg kg <sup>-1</sup>	50.0 mg L <sup>-1</sup>	±	0.2 mg L <sup>-1</sup>
Cobalt (Co)	9.47	mg kg <sup>-1</sup>	±	0.03	mg kg <sup>-1</sup>	10.00 mg L <sup>-1</sup>	±	0.03 mg L <sup>-1</sup>
Copper (Cu)	9.47	mg kg <sup>-1</sup>	±	0.03	mg kg <sup>-1</sup>	10.00 mg L <sup>-1</sup>	±	0.03 mg L <sup>-1</sup>
Iron (Fe)	9.47	mg kg <sup>-1</sup>	±	0.03	mg kg <sup>-1</sup>	10.00 mg L <sup>-1</sup>	±	0.03 mg L <sup>-1</sup>
Lead (Pb)	94.7	mg kg <sup>-1</sup>	±	0.3	mg kg <sup>-1</sup>	100.0 mg L <sup>-1</sup>	±	0.3 mg L <sup>-1</sup>
Lithium (Li)	47.3	mg kg <sup>-1</sup>	±	0.3	mg kg <sup>-1</sup>	50.0 mg L <sup>-1</sup>	±	0.3 mg L <sup>-1</sup>
Magnesium (Mg)	9.47	mg kg <sup>-1</sup>	±	0.04	mg kg <sup>-1</sup>	10.00 mg L <sup>-1</sup>	±	0.04 mg L <sup>-1</sup>
Manganese (Mn)	9.47	mg kg <sup>-1</sup>	±	0.04	mg kg <sup>-1</sup>	10.00 mg L <sup>-1</sup>	±	0.05 mg L <sup>-1</sup>
Molybdenum (Mo)	47.3	mg kg <sup>-1</sup>	±	0.2	mg kg <sup>-1</sup>	50.0 mg L <sup>-1</sup>	±	0.3 mg L <sup>-1</sup>

Certified value traceable to SI unit kg and uncertainty according to ISO Guide 35 [2] and Eurachem/CITAC Guide [3]						
Constituent	Certified values at 20°C and expanded uncertainty [U = k'uc; k = 2]					
Nickel (Ni)	47.3 mg kg <sup>-1</sup> ± 0.1 mg kg <sup>-1</sup>	50.0 mg L <sup>-1</sup> ± 0.1 mg L <sup>-1</sup>				
Potassium (K)	94.7 mg kg <sup>-1</sup> ± 0.3 mg kg <sup>-1</sup>	100.0 mg L <sup>-1</sup> ± 0.3 mg L <sup>-1</sup>				
Silver (Ag)	9.47 mg kg <sup>-1</sup> ± 0.03 mg kg <sup>-1</sup>	10.00 mg L <sup>-1</sup> ± 0.03 mg L <sup>-1</sup>				
Sodium (Na)	47.3 mg kg <sup>-1</sup> ± 0.1 mg kg <sup>-1</sup>	50.0 mg L <sup>-1</sup> ± 0.1 mg L <sup>-1</sup>				
Strontium (Sr)	9.47 mg kg <sup>-1</sup> ± 0.03 mg kg <sup>-1</sup>	10.00 mg L <sup>-1</sup> ± 0.03 mg L <sup>-1</sup>				
Thallium (TI)	47.3 mg kg <sup>-1</sup> ± 0.1 mg kg <sup>-1</sup>	50.0 mg L <sup>-1</sup> ± 0.1 mg L <sup>-1</sup>				
Zinc (Zn)	9.47 mg kg <sup>-1</sup> ± 0.03 mg kg <sup>-1</sup>	10.00 mg L <sup>-1</sup> ± 0.03 mg L <sup>-1</sup>				

## 1. CONCEPT OF CERTIFICATION AND TRACEABILITY STATEMENT

To guarantee top reliability of the values for this Trace**CERT**® certified reference material two independent procedures were followed. The values have to agree in the range of their uncertainties, but the value from the gravimetric preparation has been chosen as certified value [4]:

- 1. Gravimetric preparation using pure materials is a practical realization of concentration units, through conversion of mass to amount of substance [4]. If the purity of the materials is demonstrated and if contamination and loss of material is strictly prevented this approach allows highest accuracy and small uncertainties. The certified value of TraceCERT® reference materials is based on this approach and directly traceable to the SI unit kilogram. Therefore comprehensively characterized materials of high purity are used. All balances are certified by DKD and calibrated with OIML Class E2 (up to 12 kg) and F2 (up to 64 kg) weights.
- 2. The starting material is measured against a certified reference material (i.e. NIST or BAM) followed by gravimetric preparation using balances calibrated with SI-traceable weights. Consequently the value calculated by this unbroken chain of comparisons is traceable to the reference to which the starting material is compared.

### 2. CONTENT OF STARTING MATERIALS

For high purity materials (P >99.9%) the most appropriate way of purity determination is to quantify the impurities  $(w_i)$  and to subtract the sum from 100%. Impurities below the detection limit are considered with a contribution of half of the detection limit  $(DL_i)$ .

 $P = 100\% - \sum_{i} w_{i} - \sum_{j} \left(\frac{DL_{j}}{2}\right)$ 

Water containing materials were dried to absolute dryness by individual drying conditions (up to 600°C). When drying is impossible due to decomposition water was determined by high-precision KF-titration.



#### 3. TRACEABILITY MEASUREMENTS

Only internationally accepted reference materials e.g. from NIST (USA) or BAM (Germany) have been carefully selected to provide the basis for traceability to the SI unit Mole. To underpin the certified gravimetric value all traceability measurements are performed with the most accurate and precise analytical technique available. Therefore titrimetry measurement series are applied whenever possible (corrected for trace impurities). When no titrimetric technique is available, the traceability measurements are performed with another analytical technique, e.g. ICP-OES or AAS.

Reference and applied technique used for traceability measurements of the starting material:

Constituent	Starting material	Reference	Method
Aluminum (AI)	AI(NO <sub>3</sub> ) <sub>3</sub> x 9 H <sub>2</sub> O	NIST SRM 3101a	ICP-OES
Barium (Ba)	BaCO <sub>3</sub>	NIST SRM 728	Complexometric titration
Bismuth (Bi)	Bi metal	NIST SRM 728	Complexometric titration
Boron (B)	H <sub>3</sub> BO <sub>3</sub>	NIST SRM 84	Acid-base titration
Cadmium (Cd)	Cd metal	NIST SRM 728	Complexometric titration
Calcium (Ca)	CaCO <sub>3</sub>	NIST SRM 728	Complexometric titration
Chromium (Cr)	Cr(NO <sub>3</sub> ) <sub>3</sub> x 9 H <sub>2</sub> O	NIST SRM 3112a	ICP-OES
Cobalt (Co)	Co metal	NIST SRM 728	Complexometric titration
Copper (Cu)	Cu metal	NIST SRM 728	Complexometric titration
Iron (Fe)	Fe metal	NIST SRM 728	Complexometric titration
Lead (Pb)	Pb(NO <sub>3</sub> ) <sub>2</sub>	NIST SRM 728	Complexometric titration
Lithium (Li)	Li <sub>2</sub> CO <sub>3</sub>	NIST SRM 3129a	ICP-OES
Magnesium (Mg)	Mg(NO <sub>3</sub> ) <sub>3</sub> x 6 H <sub>2</sub> O	NIST SRM 728	Complexometric titration
Manganese (Mn)	Mn(NO <sub>3</sub> ) <sub>2</sub> x X H <sub>2</sub> O	NIST SRM 3132	ICP-OES
Molybdenum (Mo)	MoO <sub>3</sub>	NIST SRM 3134	ICP-OES
Nickel (Ni)	Ni metal	NIST SRM 728	Complexometric titration
Potassium (K)	KNO <sub>3</sub>	NIST SRM 3141a	ICP-OES
Silver (Ag)	Ag metal	NIST SRM 723	Argentometric titration
Sodium (Na)	NaNO <sub>3</sub>	NIST SRM 3152a	ICP-OES
Strontium (Sr)	Sr(NO <sub>3</sub> ) <sub>2</sub>	NIST SRM 728	Complexometric titration
Thallium (TI)	TINO <sub>3</sub>	NIST SRM 3158	Redox titration
Zinc (Zn)	Zn metal	NIST SRM 728	Complexometric titration

#### 4. STORING BEHAVIOR

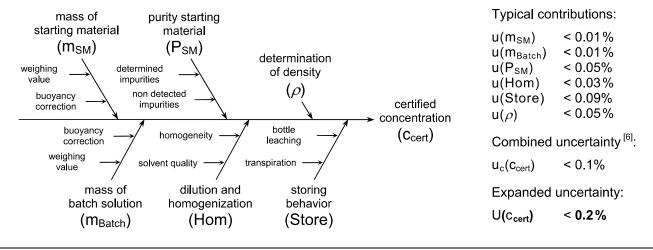
The storage behavior of standard solutions is of greatest importance with regard to the certified value. Therefore the two most important effects were investigated by in-depth studies in a cooperation with EMPA, St. Gallen:

- 1. The leach out of trace impurities from HDPE (high-density polyethylene) bottles was determined with HR-ICP-MS after leaching the bottles with 2% nitric acid. Maximum contamination levels were found in the ng L<sup>-1</sup> level for 12 elements.
- 2. To avoid significant loss of mass through transpiration the bottle is delivered in aluminum coated bags. After the bottle has been removed from the bag, transpiration will occur at an accelerated rate (see figure). We highly recommend not to open the bag until the solution is needed. Once the bottle is opened the solution should be stored at reduced temperature (4°C) to reduce transpiration.

Transpiration through 100 mL HD-PE bottles 1.008 measured (mean of 9 bottles) extrapolated 30°C Increase of concentratior 1.006 relative to 1 000 gL<sup>-1</sup> Bottle stored without Al-bag 1.004 23°C 1.002 Bottle stored in Al-bag, 23°C 1.000 2 3 Years

#### 5. Uncertainty Evaluation

All uncertainties are calculated according to Eurachem/CITAC Guide [3] and reported as combined expanded uncertainties at the 95% confidence level. For gravimetric preparation the uncertainty contributions are illustrated by the following cause-effect diagram [5]:



**CRM** operations: S.Matt Approving Officer: **ISO 17034 ISO/IEC 17025 ISO 9001** P. Zell, Ph.D. **SRMS 0001** 

- ISO Guide 31:2015, "Reference materials Contents of certificates, labels and accompanying documentation"
- [2] ISO Guide 35:2017, "Reference materials - Guidance for characterization and assessment of homogeneity and stability"
- Eurachem/CITAC Guide, 3<sup>rd</sup> Ed. (2012), "Quantifying uncertainty in analytical measurement" Eurachem/CITAC Guide, 1<sup>st</sup> Ed. (2003) "Traceability in chemical measurement" [3] [4]
- Reichmuth, A., Wunderli, S., Weber, M., Meyer, V. R. (2004), "The uncertainty of weighing data obtained with electronic analytical balances", [5] Microchimica Acta 148: 133-141.
- [6] Calculated by combination of the squared contribution values

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